

**Characterizing the Invisible:
EPR and ENDOR Studies of Intermediates in N₂ Reduction by Nitrogenase**

Hong-In Lee[&], Robert Y. Igarashi[§], Mikhail Laryukhin[#], Patricia C. Dos Santos[¶],
Tran-Chin Yang[#], Dennis R. Dean[¶], Lance C. Seefeldt[§], and Brian M. Hoffman^{#*}

[&]Department of Chemistry Education, Kyungpook National University,
Daegu 702-701, Korea

[§]Department of Chemistry and Biochemistry, Utah State University, Logan, Utah
84322, USA

[#]Department of Chemistry, Northwestern University, Evanston, Illinois 60208,
USA

[¶]Department of Biochemistry, Virginia Tech, Blacksburg, Virginia 24061, USA

Characterization of an enzyme mechanism can be viewed as the process of identifying and characterizing the structure and reactivity of intermediates that comprise its catalytic cycle. This talk will illustrate our approach to trapping such intermediates and using electron paramagnetic resonance (EPR) and electron-nuclear double resonance (ENDOR) spectroscopies to characterize them, through discussion of nitrogenase. Nitrogenase catalyzes the reduction of N₂ to NH₃ on the most complicated metalloenzyme active site: the molybdenum-iron cofactor (FeMo-cofactor), an [Fe₇MoS₉X] cluster. Despite the availability of the beautiful and mysterious structure of this cofactor for more than a decade, it has yielded no clue as to the site(s) of substrate binding and reaction. We have now: (i) determined the site of alkyne and dinitrogen reduction, (ii) characterized an intermediate trapped during the reduction of propargyl alcohol ((HC≡C-CH₂OH; PA) through use of a novel set of *quantitative* ^{1,2}H ENDOR spectroscopic techniques, plus ¹³C ENDOR measurements; (iii) used ^{1,2}H ENDOR to study an intermediate that is formed during proton reduction and has two H^{+/-} bound to the cofactor; (iii) are in the process of characterizing *several* reduced forms of N₂ bound to the cofactor.