

Solvent Effects on the Energetics of Prolyl Peptide Bond Isomerization

Supplementary Material

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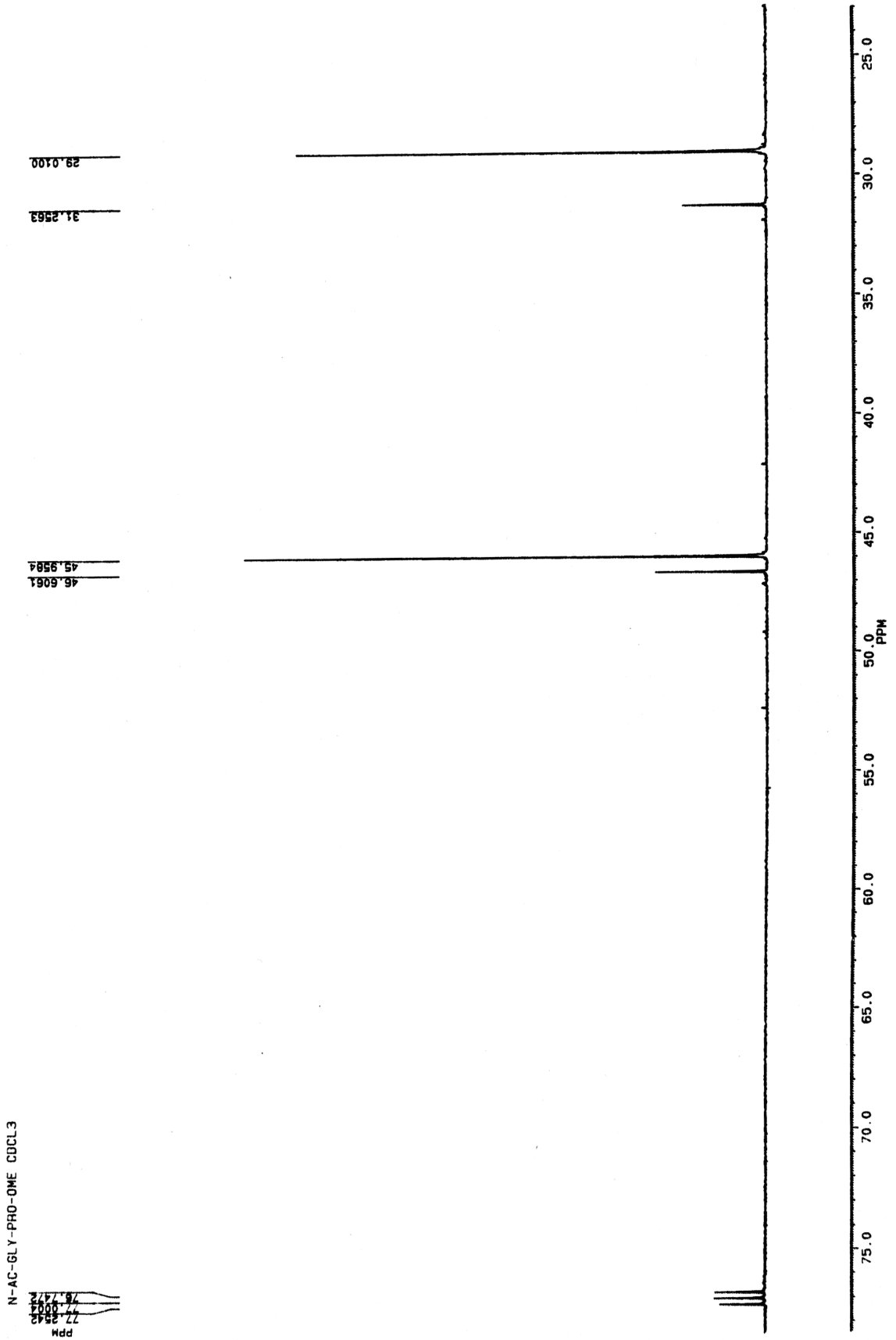
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Figure Legends

Figure 1S. ^{13}C NMR spectrum (CDCl_3 , 125.77 MHz) of **1**. δ (in ppm): 29.0100 (Pro β , *trans*), 31.2563 (Pro β , *cis*), 45.9584 (Pro δ , *trans*), 46.6061 (Pro δ , *cis*). Pro β was used for inversion transfer experiments. FAB-MS of **1** showed $(\text{M} + \text{H})^+ = 231$ u (calculated 231.2 u).

Figure 2S. FT-IR spectrum [100 mM sodium phosphate buffer, pH 7.2, containing 20% (v/v) D_2O ; ZnSe crystal] of Ac-Pro-OMe. ν (in cm^{-1}): 1605.95 (amide), 1729.50 (ester).



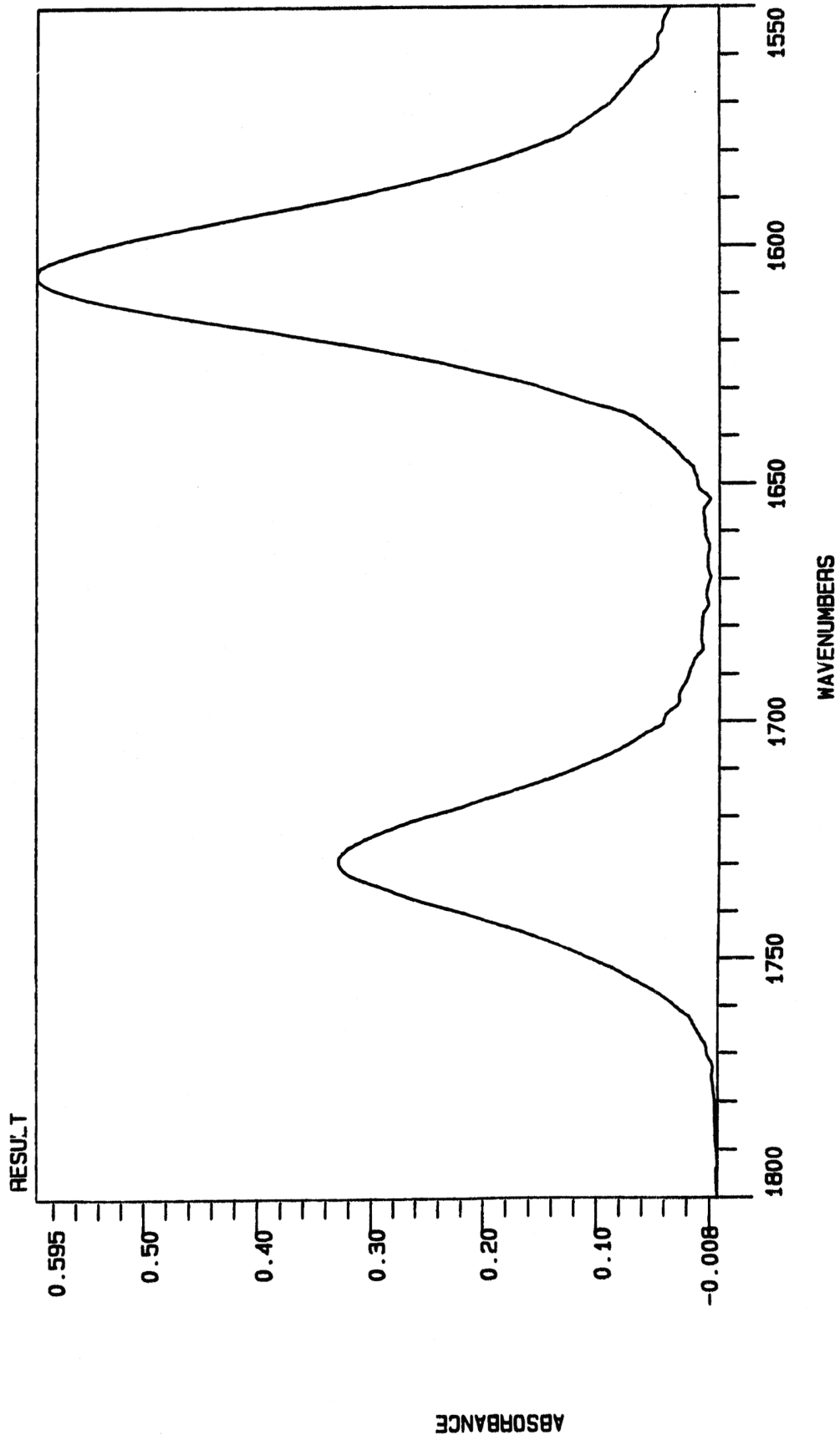


Table IS. Solvent Effects on Activation Parameters for Isomerization of **1**.

solvent	rate constant	ΔH^\ddagger ^a [kcal/mol]	ΔS^\ddagger ^a [cal/(mol K)]
acetonitrile	k_{EZ}	19.25 ± 0.14	3.18 ± 0.43
	k_{ZE}	19.97 ± 0.14	2.71 ± 0.43
benzene	k_{EZ}	20.79 ± 0.29	7.74 ± 0.87
	k_{ZE}	19.46 ± 0.09	0.66 ± 0.27
<i>N,N</i> -dimethylformamide	k_{EZ}	17.55 ± 0.14	-2.29 ± 0.41
	k_{ZE}	18.62 ± 0.05	-0.73 ± 0.15
dioxane	k_{EZ}	16.67 ± 0.03	-3.84 ± 0.08
	k_{ZE}	18.30 ± 0.09	-1.85 ± 0.26
ethanol	k_{EZ}	15.35 ± 0.29	-9.61 ± 0.88
	k_{ZE}	14.96 ± 0.10	-13.59 ± 0.32
isopropanol	k_{EZ}	17.68 ± 0.29	-2.01 ± 0.67
	k_{ZE}	19.42 ± 0.16	0.69 ± 1.14
toluene	k_{EZ}	22.28 ± 0.11	12.95 ± 0.34
	k_{ZE}	23.35 ± 0.06	13.86 ± 0.20
trifluoroethanol	k_{EZ}	20.60 ± 0.29	2.97 ± 0.86
	k_{ZE}	23.08 ± 0.13	6.89 ± 0.40
water	k_{EZ}	18.78 ± 0.13	-1.42 ± 0.39
	k_{ZE}	21.21 ± 0.13	2.29 ± 0.38

^aValues were calculated from linear regression analysis of plots of $\ln(k_{EZ}/T)$ vs $1/T$ and $\ln(k_{ZE}/T)$ vs $1/T$.