

Biosynthesis of Nucleoside Diphosphoramidates in *Campylobacter jejuni*

Zane W. Taylor,[†] Haley A. Brown,[§] Hazel M. Holden,^{*,§} and Frank M. Raushel^{*,†,‡,§}

[†]Department of Biochemistry & Biophysics, Texas A&M University, College Station, Texas 77843, United States

[‡]Department of Chemistry, Texas A&M University, College Station, Texas 77843, United States

[§]Department of Biochemistry, University of Wisconsin—Madison, Madison, Wisconsin 53706, United States

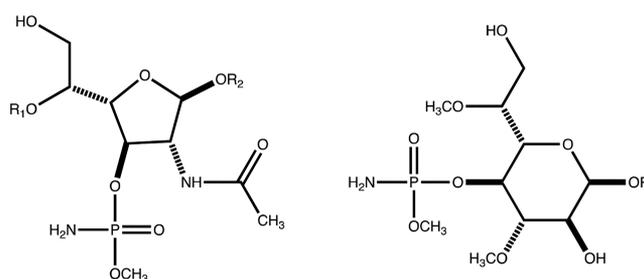
Supporting Information

ABSTRACT: *Campylobacter jejuni* is a pathogenic Gram-negative bacterium and a leading cause of food-borne gastroenteritis. *C. jejuni* produces a capsular polysaccharide (CPS) that contains a unique *O*-methyl phosphoramidate modification (MeOPN). Recently, the first step in the biosynthetic pathway for the assembly of the MeOPN modification to the CPS was elucidated. It was shown that the enzyme Cj1418 catalyzes the phosphorylation of the amide nitrogen of *L*-glutamine to form *L*-glutamine phosphate. In this investigation, the metabolic fate of *L*-glutamine phosphate was determined. The enzyme Cj1416 catalyzes the displacement of pyrophosphate from MgCTP by *L*-glutamine phosphate to form CDP-*L*-glutamine. The enzyme Cj1417 subsequently catalyzes the hydrolysis of CDP-*L*-glutamine to generate cytidine diphosphoramidate and *L*-glutamate. The structures of the two novel intermediates, CDP-*L*-glutamine and cytidine diphosphoramidate, were confirmed by ³¹P nuclear magnetic resonance spectroscopy and mass spectrometry. It is proposed that the enzyme Cj1416 be named CTP:phosphoglutamine cytidyltransferase and that the enzyme Cj1417 be named γ -glutamyl-CDP-amidate hydrolase.

The pathogenic Gram-negative bacterium *Campylobacter jejuni* is a leading cause of food-borne gastroenteritis.¹ While pathogenic to humans, *C. jejuni* is a commensal organism in chickens, and as a result, contaminated poultry serves as a common route of human infection. While most *C. jejuni* infections cause a case of gastroenteritis, approximately 1 in 1000 infections results in the autoimmune disease Guillain-Barré syndrome.^{2,3} Like many other organisms, *C. jejuni* uses a capsular polysaccharide (CPS) to improve fitness. The capsular polysaccharide of *C. jejuni* protects the organism from bacteriophages and shields it from the host immune response.^{4,5} More than 40 different strains of *C. jejuni* have been identified to date, and each strain is believed to produce a unique CPS variant.^{6,7} In *C. jejuni* strain NCTC 11168, a cluster of approximately 35 genes is responsible for the synthesis and export of the CPS.⁸ Moreover, *C. jejuni* has evolved the ability to synthesize a unique *O*-methyl phosphoramidate (MeOPN) modification found on the CPS that improves the pathogenicity of the bacterium and promotes evasion of the host immune response.⁹ The structures of the MeOPN modification to the

CPS in *C. jejuni* strain NCTC 11168 and hypermotile strain 11168H are illustrated in Scheme 1.

Scheme 1. Structures of the *O*-Methyl Phosphoramidate Modifications to the CPS in *C. jejuni*



We have recently shown that the enzyme denoted with the locus tag Cj1418 from *C. jejuni* strain 11168H catalyzes the first committed step in the biosynthesis of the MeOPN.¹⁰ This enzyme, *L*-glutamine kinase, catalyzes the unprecedented ATP-dependent phosphorylation of the amide nitrogen of *L*-glutamine (1) to form *L*-glutamine phosphate (2) as shown in Scheme 2A. However, the subsequent metabolic fate of this novel enzyme intermediate has not been addressed. The primary focus of this investigation is to identify those enzymes that can harness the phosphoramidate moiety contained within *L*-glutamine phosphate for the ultimate construction of the *O*-methyl phosphoramidate modification of the CPS.

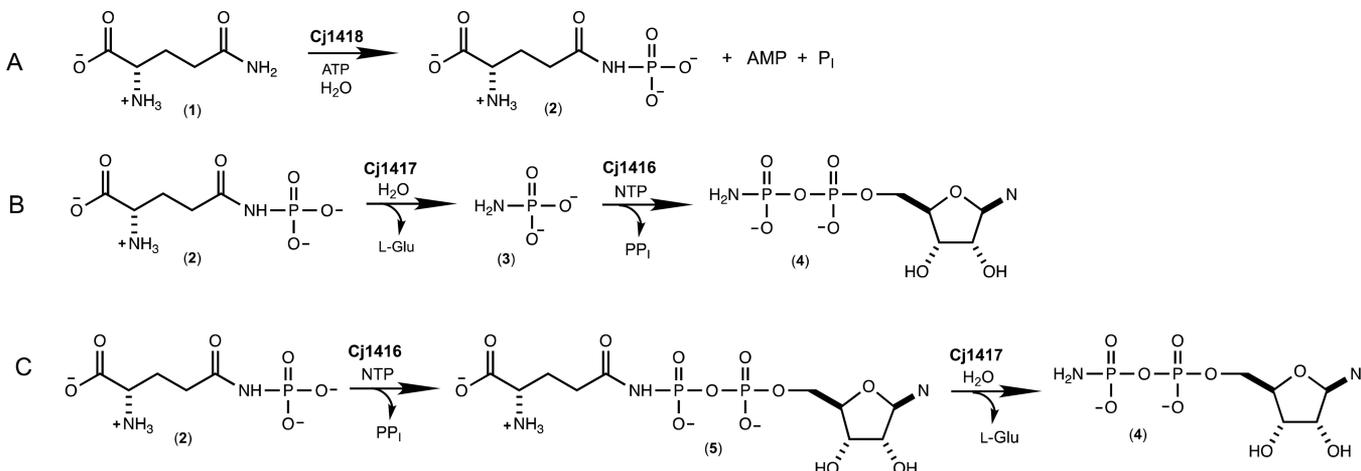
The two most likely enzymes of unknown function from *C. jejuni* to utilize *L*-glutamine phosphate as a substrate during the biosynthesis of the *O*-methyl phosphoramidate modification of the CPS are Cj1417 and Cj1416. Cj1417 is functionally annotated as a Type I amidotransferase from cog2071. Members of the Type I amidotransferase family of enzymes typically catalyze the hydrolysis of *L*-glutamine or amide-substituted derivatives of this amino acid.^{11,12} Currently, the closest functionally characterized homologue to Cj1417 is the enzyme PuuD (23% identical sequence) from *Escherichia coli*, an enzyme that catalyzes the hydrolysis of 4-(γ -*L*-glutamylamino)butanoate to 4-aminobutanoate and *L*-glutamate.¹² Cj1416 is currently annotated as a nucleotidyl

Received: September 12, 2017

Revised: October 11, 2017

Published: October 12, 2017

Scheme 2. Reaction Catalyzed by Cj1418 and the Predicted Functions of Cj1417 and Cj1416



transferase from cog1213. The closest functionally characterized homologue of Cj1416 is CTP:phosphocholine cytidyltransferase from *Streptococcus pneumoniae* with a sequence identity of 28%.¹³ This enzyme catalyzes the formation of CDP-choline and pyrophosphate from CTP and choline phosphate. We therefore predict that the combination of Cj1417 and Cj1416 will catalyze the synthesis of a nucleoside diphosphoramidate (4) using L-glutamine phosphate (2) and a nucleoside triphosphate as substrates.

The biosynthesis of a nucleoside diphosphoramidate (4) by the consecutive reactions catalyzed by Cj1417 and Cj1416 can be envisioned to occur via one of two possible reaction schemes. In the first scenario, Cj1417 catalyzes the hydrolysis of L-glutamine phosphate (2) to L-glutamate and phosphoramidate (3). This reaction is followed by the displacement of pyrophosphate from a nucleoside triphosphate by phosphoramidate (3) to generate the nucleoside diphosphoramidate (4) in a reaction catalyzed by Cj1416 as illustrated in Scheme 2B. Alternatively, Cj1416 catalyzes the displacement of pyrophosphate from a nucleoside triphosphate by L-glutamine phosphate to form NDP-L-glutamine (5). This reaction is subsequently followed by the hydrolysis of this intermediate by Cj1417 to form L-glutamate and the nucleoside diphosphoramidate (4) as presented in Scheme 2C.

To test our initial prediction that the combination of Cj1416 and Cj1417 catalyzes the formation of a nucleoside diphosphoramidate, these two enzymes were incubated together in the presence of MgCTP and an excess of L-glutamine phosphate (2) at pH 8.0. After 45 min, all of the CTP [retention time of 8.7 min (Figure 1A)] was converted to a new product with a retention time of 5.9 min (Figure 1B). The retention time of the new reaction product formed in the presence of Cj1417 and Cj1416 is identical to that of authentic cytidine diphosphoramidate (Figure 1E).¹⁴

The identity of the new reaction product as cytidine diphosphoramidate [4 (Scheme 2C)] was confirmed by ³¹P nuclear magnetic resonance (NMR) spectroscopy. A reaction mixture containing CTP, MgCl₂, and an excess of L-glutamine phosphate (2) was incubated at pH 8.0 for 90 min in the presence of Cj1416 and Cj1417 until the reaction was quenched by the addition of 10 mM EDTA. The ³¹P NMR spectrum of the control reaction (Figure 2A), obtained in the absence of Cj1416 and Cj1417, showed the expected resonances for CTP [−20.99 ppm (β-P), −10.33 ppm (α-P),

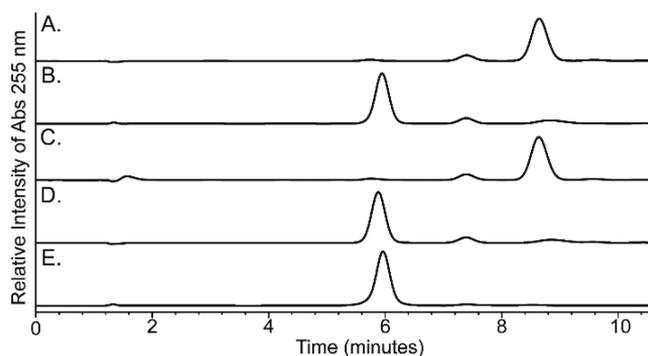


Figure 1. Anion exchange chromatograms of nucleotide standards and enzyme-catalyzed reaction products formed in 100 mM HEPES (pH 8.0) at 25 °C with an incubation time of 45 min. The elution profiles were monitored at 255 nm. The nucleotides were separated using a 0 to 17% gradient of 10 mM triethanolamine (pH 8) and 2 M KCl over 17 column volumes on a 1 mL Resource Q column. (A) Control sample of 1.0 mM CTP and 2.0 mM MgCl₂ in the absence of any added enzyme. (B) Sample containing 1.0 mM CTP, 2.0 mM MgCl₂, 5.0 mM L-glutamine phosphate, Cj1416 (5 μM), and Cj1417 (5 μM). (C) Sample containing 1.0 mM CTP, 2.0 mM MgCl₂, 5.0 mM phosphoramidate (3), 5 μM Cj1416, and 5 units/mL pyrophosphatase. (D) Sample containing 1.0 mM CTP, 2.0 mM MgCl₂, 5.0 mM L-glutamine phosphate, 5 μM Cj1416, and 5 units/mL pyrophosphatase. (E) Control sample of chemically synthesized CDP phosphoramidate.

and −5.51 ppm (γ-P)] and L-glutamine phosphate (−3.57 and −3.83 ppm). In the presence of Cj1416 and Cj1417, the resonances for CTP and L-glutamine phosphate (2) essentially disappear and are replaced by new resonances for pyrophosphate (−6.63 ppm) and a pair of doublets at −0.42 ppm (β-P) and −10.23 ppm (α-P) for cytidine diphosphoramidate (Figure 2B). The ³¹P NMR spectrum for authentic cytidine diphosphoramidate is presented in Figure 2C. The formation of cytidine diphosphoramidate was further supported by the acquisition of the negative ion electrospray ionization (ESI) mass spectrum of the unfractionated reaction mixture upon incubation of MgCTP, L-glutamine phosphate, Cj1417, and Cj1416 at pH 8.0 in ammonium bicarbonate buffer. A peak at *m/z* 401.03 was observed that is consistent with that expected for cytidine diphosphoramidate (Figure S1). These experiments demonstrate that Cj1417 and Cj1416 use MgCTP and L-glutamine phosphate (2) to catalyze the formation of CDP phosphoramidate, pyrophosphate, and L-glutamate.

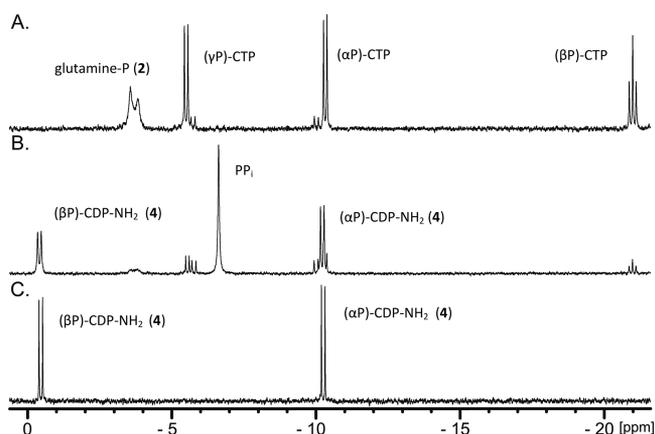


Figure 2. ^{31}P NMR spectra of nucleotide standards and enzyme-catalyzed reaction products formed in 100 mM HEPES (pH 8.0) at 25 °C with an incubation time of 90 min before the reaction was quenched with 10 mM EDTA. (A) Control sample containing 5.0 mM CTP, 5.0 mM MgCl_2 , and 6.0 mM L-glutamine phosphate. (B) Sample containing 5.0 mM CTP, 5.0 mM MgCl_2 , 6.0 mM L-glutamine phosphate, 20 μM Cj1416, and 20 μM Cj1417. (C) Control sample of 5.0 mM CDP phosphoramidate.

In parts B and C of Scheme 2, we have proposed that either phosphoramidate (3) or L-glutamine phosphate (2) is used to displace pyrophosphate from a nucleoside triphosphate to form either a nucleoside diphosphoramidate (4) or NDP-L-glutamine (5) as an intermediate. The reactivity of Cj1416 with each of these potential substrates was tested with MgCTP as the acceptor nucleotide, and the reaction was monitored by ion exchange chromatography at 255 nm. In separate experiments, either phosphoramidate (3) or L-glutamine phosphate (2) was incubated with CTP, MgCl_2 , and Cj1416 at pH 8.0 for 45 min. Utilizing the chemically synthesized phosphoramidate (3) as a potential substrate, there was no change in the high-performance liquid chromatography chromatogram when compared to that of CTP alone (Figure 1C).¹⁵ However, when Cj1416 was incubated with L-glutamine phosphate (2) and MgCTP, all of the CTP was converted to a new product that corresponds to a molecule with a net charge of approximately -2 (Figure 1D).¹⁰ This result demonstrates that Cj1416 is fully capable of using L-glutamine phosphate (2) to displace pyrophosphate from CTP.

To provide further spectroscopic support for the Cj1416-catalyzed formation of CDP-L-glutamine, the reaction products were analyzed by ^{31}P NMR spectroscopy. In this experiment, Cj1416 was incubated with CTP, MgCl_2 , L-glutamine phosphate (2), and inorganic pyrophosphatase until the reaction was quenched with EDTA. After an incubation period of 90 min, essentially all of the CTP and L-glutamine phosphate (2) were converted to products (Figure 3A). A new resonance is observed at 3.02 ppm for phosphate (from the hydrolysis of pyrophosphate), and two new doublets are observed at -10.98 ppm ($\alpha\text{-P}$) and -16.13 ppm ($\beta\text{-P}$) for CDP-L-glutamine. The formation of CDP-L-glutamine was further supported by the acquisition of the negative ion ESI mass spectrum of the unfractionated reaction mixture upon incubation of MgCTP, L-glutamine phosphate, and Cj1416 at pH 8.0 in ammonium bicarbonate buffer. A peak at m/z 530.07 was observed that is consistent with that expected of CDP-L-glutamine (Figure S2).

When Cj1417 is subsequently added to the reaction mixture containing CDP-L-glutamine, the ^{31}P NMR resonances for

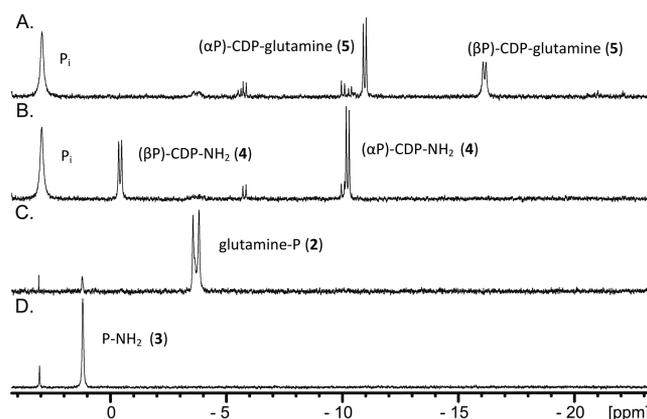


Figure 3. ^{31}P NMR spectra of nucleotide standards and enzyme-catalyzed reaction products formed in 100 mM HEPES (pH 8.0) at 25 °C with an incubation time of 90 min before the reaction was quenched with 10 mM EDTA. (A) Sample containing 5.0 mM CTP, 5.0 mM MgCl_2 , 5 units/mL pyrophosphatase, and 20 μM Cj1416. (B) Cj1417 (20 μM) was added to the reaction mixture shown in panel A and the mixture allowed to react for an additional 90 min. (C) Sample containing 20 μM Cj1417 and 5.0 mM glutamine phosphate. (D) Control sample of 5.0 mM phosphoramidate.

CDP-L-glutamine disappear and are replaced by two new pairs of doublets at -0.43 ppm ($\beta\text{-P}$) and -10.24 ppm ($\alpha\text{-P}$) that can be assigned to cytidine diphosphoramidate (Figure 3B and Figure 2C). These experiments demonstrate that Cj1416 catalyzes the formation of CDP-L-glutamine from CTP and L-glutamine phosphate (2) and that Cj1417 catalyzes the hydrolysis of CDP-L-glutamine to L-glutamate and cytidine diphosphoramidate.

The remaining issue to address for the functional characterization of Cj1417 is whether this enzyme is capable of catalyzing the hydrolysis of L-glutamine phosphate to L-glutamate and phosphoramidate (2). Cj1417 was therefore incubated with L-glutamine phosphate (2) at pH 8.0 for 90 min. The ^{31}P NMR spectrum (Figure 3C) of the reaction mixture demonstrated that $\sim 93\%$ of the original L-glutamine phosphate (2) remained intact. Two other resonances that are consistent with the presence of a small amount of phosphate (3.08 ppm, 1.2%) and phosphoramidate (1.21 ppm, 5.5%) are observed. The ^{31}P NMR spectrum for chemically synthesized phosphoramidate (2) is shown in Figure 3D, where a resonance is observed at 1.19 ppm. On the basis of these results, it is clear that the preferred pathway for the synthesis of cytidine diphosphoramidate (4) is for Cj1416 to catalyze the displacement of pyrophosphate from CTP to form CDP-L-glutamine (5) and for Cj1417 to catalyze the hydrolysis of this intermediate to generate cytidine diphosphoramidate (4) as shown in Scheme 2C.

The kinetic constants for the catalytic activity of Cj1417 and Cj1416 were determined. At a fixed concentration of either 5.0 mM MgCTP or 2.0 mM L-glutamine phosphate (2), the rates of the reaction catalyzed by Cj1416 were determined at 25 °C and pH 8.0 by monitoring the formation of products via anion exchange chromatography at 255 nm. The observed kinetic constants using L-glutamine phosphate (2) as the variable substrate are as follows: $K_m = 120 \pm 30 \mu\text{M}$, $k_{\text{cat}} = 57 \pm 6 \text{ min}^{-1}$, and $k_{\text{cat}}/K_m = (4.8 \pm 1.3) \times 10^5 \text{ M}^{-1} \text{ min}^{-1}$. The observed kinetic constants using MgCTP as the variable substrate are as follows: $K_m = 170 \pm 35 \mu\text{M}$, $k_{\text{cat}} = 57 \pm 6 \text{ min}^{-1}$, and $k_{\text{cat}}/K_m = (3.4 \pm 0.8) \times 10^5 \text{ M}^{-1} \text{ min}^{-1}$.

The kinetic constants for the hydrolysis of CDP-L-glutamine (5) catalyzed by Cj1417 were determined at 25 °C and pH 8.0 using a glutamate dehydrogenase coupled assay that monitors the formation of NADH at 340 nm. The kinetic constants were determined to be as follows: $K_m = 28 \pm 3 \mu\text{M}$, $k_{\text{cat}} = 34 \pm 1.2 \text{ min}^{-1}$, and $k_{\text{cat}}/K_m = (1.2 \pm 0.2) \times 10^6 \text{ M}^{-1} \text{ min}^{-1}$. In an effort to determine the upper limit of the rate constant for the hydrolysis of L-glutamine phosphate (2) by Cj1417, 10 mM L-glutamine phosphate (2) was incubated with 20 μM Cj1417, and the reaction was monitored by ^{31}P NMR for 13 h. From this experiment, an upper limit of 1.6 h^{-1} was obtained for the hydrolysis of L-glutamine phosphate by Cj1417.

Previously, we have demonstrated that the first step in the biosynthesis of the MeOPN modification to the CPS of *C. jejuni* is catalyzed by Cj1418, where ATP is utilized to phosphorylate the amide nitrogen of L-glutamine.¹⁰ Here we have shown that Cj1416 catalyzes the displacement of pyrophosphate from MgCTP by L-glutamine phosphate (2), yielding CDP-L-glutamine (5). We have also established that the catalytic function of Cj1417 is to hydrolyze CDP-L-glutamine to L-glutamate and cytidine diphosphoramidate (4). This investigation has thus unveiled the identities of two new nucleoside diphosphoramidate derivatives that are involved in the biosynthesis of MeOPN. It is likely that cytidine diphosphoramidate (4) will be subsequently phosphorylated at the hydroxyl group attached to C3 of the ribose ring prior to transfer of the phosphoramidate group to various carbohydrates of the CPS. On the basis of sequence similarity network analysis of the enzymes of unknown function contained within *C. jejuni*, we predict that Cj1415 will catalyze this reaction. Cj1415 is a member of cog0529, and the closest functionally characterized enzyme is CysC, an adenylyl-sulfate kinase, from *E. coli* (26% identity), an enzyme that catalyzes the ATP-dependent phosphorylation of the 3'-hydroxyl of adenylyl sulfate.¹⁶ The unique biosynthetic pathway for the assembly of the phosphoramidate functionality found in the CPS of *C. jejuni* offers many opportunities for the development of potent inhibitors that may ultimately be useful in the therapeutic control of this pathogenic organism.

■ ASSOCIATED CONTENT

■ Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.biochem.7b00905.

Methods, Figures S1 and S2, and additional references (PDF)

■ AUTHOR INFORMATION

Corresponding Authors

*E-mail: raushel@tamu.edu.

*E-mail: hmholden@wisc.edu.

ORCID

Frank M. Raushel: 0000-0002-5918-3089

Funding

This work was supported in part by grants from the Robert A. Welch Foundation (A-840) to F.M.R. and the National Institutes of Health (GM 115921) to H.M.H.

Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

We thank Professor C. Szymanski and Mr. C. Wenzel for the plasmids that were used to express Cj1417 and Cj1416.

■ REFERENCES

- (1) Young, K. T., Davis, L. M., and Dirita, V. J. (2007) *Nat. Rev. Microbiol.* 5, 665–679.
- (2) Jacobs, B. C., Rothbarth, P. H., van der Meché, F. G. A., Herbrink, P., Schmitz, P. I. M., de Klerk, M. A., and van Doorn, P. A. (1998) *Neurology* 51, 1110–1115.
- (3) Acheson, D., and Allos, B. M. (2001) *Clin. Infect. Dis.* 32, 1201–1206.
- (4) Guerry, P., Poly, F., Riddle, M., Maue, A., Chen, Y.-H., and Monteiro, M. A. (2012) *Front. Cell. Infect. Microbiol.* 2, 1–7.
- (5) Sørensen, M. C. H., van Alphen, L. B., Harboe, A., Li, J., Christensen, B. B., Szymanski, C. M., and Brøndsted, L. (2011) *J. Bacteriol.* 193, 6742–6749.
- (6) Karlyshev, A. V., Champier, O. L., Churcher, C., Brisson, J.-R., Jarrell, H. C., Gilbert, M., Brochu, D., St Michael, F., Li, J., Wakarchuk, W. W., Goodhead, I., Sanders, M., Stevens, K., White, B., Parkhill, J., Wren, B. W., and Szymanski, C. M. (2005) *Mol. Microbiol.* 55, 90–103.
- (7) Michael, F. S., Szymanski, C. M., Li, J., Chan, K. H., Khieu, N. H., Larocque, S., Wakarchuk, W. W., Brisson, J.-R., and Monteiro, M. A. (2002) *Eur. J. Biochem.* 269, 5119–5136.
- (8) Karlyshev, A. V., Linton, D., Gregson, N. A., Lastovica, A. J., and Wren, B. W. (2000) *Mol. Microbiol.* 35, 529–541.
- (9) van Alphen, L. B., Wenzel, C. Q., Richards, M. R., Fodor, C., Ashmus, R. A., Stahl, M., Karlyshev, A. V., Wren, B. W., Stintzi, A., Miller, W. G., Lowary, T. L., and Szymanski, C. M. (2014) *PLoS One* 9, e87051.
- (10) Taylor, Z. W., Brown, H. A., Narindoshvili, T., Wenzel, C. Q., Szymanski, C. M., Holden, H. M., and Raushel, F. M. (2017) *J. Am. Chem. Soc.* 139, 9463–9466.
- (11) Massière, F., and Badet-Denisot, M.-A. (1998) *Cell. Mol. Life Sci.* 54, 205–222.
- (12) Kurihara, S., Oda, S., Kato, K., Kim, H. G., Koyanagi, T., Kumagai, H., and Suzuki, H. (2005) *J. Biol. Chem.* 280, 4602–4608.
- (13) Kwak, B.-Y., Zhang, Y.-M., Yun, M., Heath, R. J., Rock, C. O., Jackowski, S., and Park, H.-W. (2002) *J. Biol. Chem.* 277, 4343–4350.
- (14) Wehrli, W., Verheyden, D., and Moffatt, J. (1965) *J. Am. Chem. Soc.* 87, 2265–2277.
- (15) Watanabe, M., Sato, S., and Wakasugi, K. (1990) *Bull. Chem. Soc. Jpn.* 63, 1243–1245.
- (16) Leyh, T. S., Taylor, J. C., and Markham, G. D. (1988) *J. Biol. Chem.* 263, 2409–2416.